Name:		
Hour:_	Date:_	

Types of Chemical Bonds Jig-Saw

<u>Wh</u>	y A	Atoms Bond (p.40-41)
	1.) What is a compound?
	2.) Why do elements form compounds?
	3.) In what way does bonding center around electrons?
	4.) Where can you find the most stable elements on the periodic table? Give an example of one of these elements. (p. 36)
	5.) When is an atom likely to for a chemical bond?
	6.) List the three types of chemical bonds.
<u>Туг</u>	oes	of Chemical Bonds
	7.) What charge does an electron have?
		What charge does a proton have?
	8.) When an atom gains an electron, what charge does that atom have? Explain why.
	9.) When an atom loses an electron, what charge does that atom have? Explain why.
	10.) What is an ion?
	11.) What is an ionic bond?

Covalent Bonds

12	What	oro	001/0	lant	bonds?
14.	vv Hat	are	Cova	lem	DOMUS (

13. What is a molecule? Give an example of	e? Give an example of one.
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Metallic Bonds

- 14. Describe three properties of metals.
 - a.
 - b.
 - c.
- 15. What are metallic bonds?